## **Abstract**

In this work, Mn<sub>2</sub>O<sub>3</sub> were synthesized by molten salt method and WO<sub>3</sub> via hydrothermal treatment. Metal oxide-polymer based composites were prepared by incorporating the natural biopolymer, chitosan that is environment friendly and biodegradable. Direct blending strategy was applied to make Mn<sub>2</sub>O<sub>3</sub>, WO<sub>3</sub> and chitosan hybrid composites. Drop casting method was utilized for facile fabrication of composite films on FTO glass substrate. FTIR analysis exhibited the functional groups of chitosan in hybrid composite films. XRD and Raman analysis described structural properties pristine Mn<sub>2</sub>O<sub>3</sub> and WO<sub>3</sub>, (Mn<sub>2</sub>O<sub>3</sub>- chitosan), (WO<sub>3</sub>-chitosan) and (Mn<sub>2</sub>O<sub>3</sub>-WO<sub>3</sub>-chitosan) confirmed the formation of cubic crystalline Mn<sub>2</sub>O<sub>3</sub> and hexagonal WO<sub>3</sub>. Thermal stability of Mn<sub>2</sub>O<sub>3</sub> -WO<sub>3</sub> -chitosan composite was investigated by Thermogravimetric Analysis/Differential Scanning Calorimetry (TGA/DSC) that depicted two stages of degradation, first one attributed to the removal of surface water and second to the decomposition of amine units. Scanning Electron Microscope images revealed the compact granular surface morphology of (Mn<sub>2</sub>O<sub>3</sub> -WO<sub>3</sub> -chitosan) composite depicting stronger interaction of metal oxides and chitosan. Optical studies demonstrated that band gap of, (Mn<sub>2</sub>O<sub>3</sub>- chitosan), (WO<sub>3</sub>-chitosan) and (Mn<sub>2</sub>O<sub>3</sub>-WO<sub>3</sub>-chitosan) composite films were 2.13 eV, 2.43 eV, and 2.48 eV respectively and their photodegradation studies.